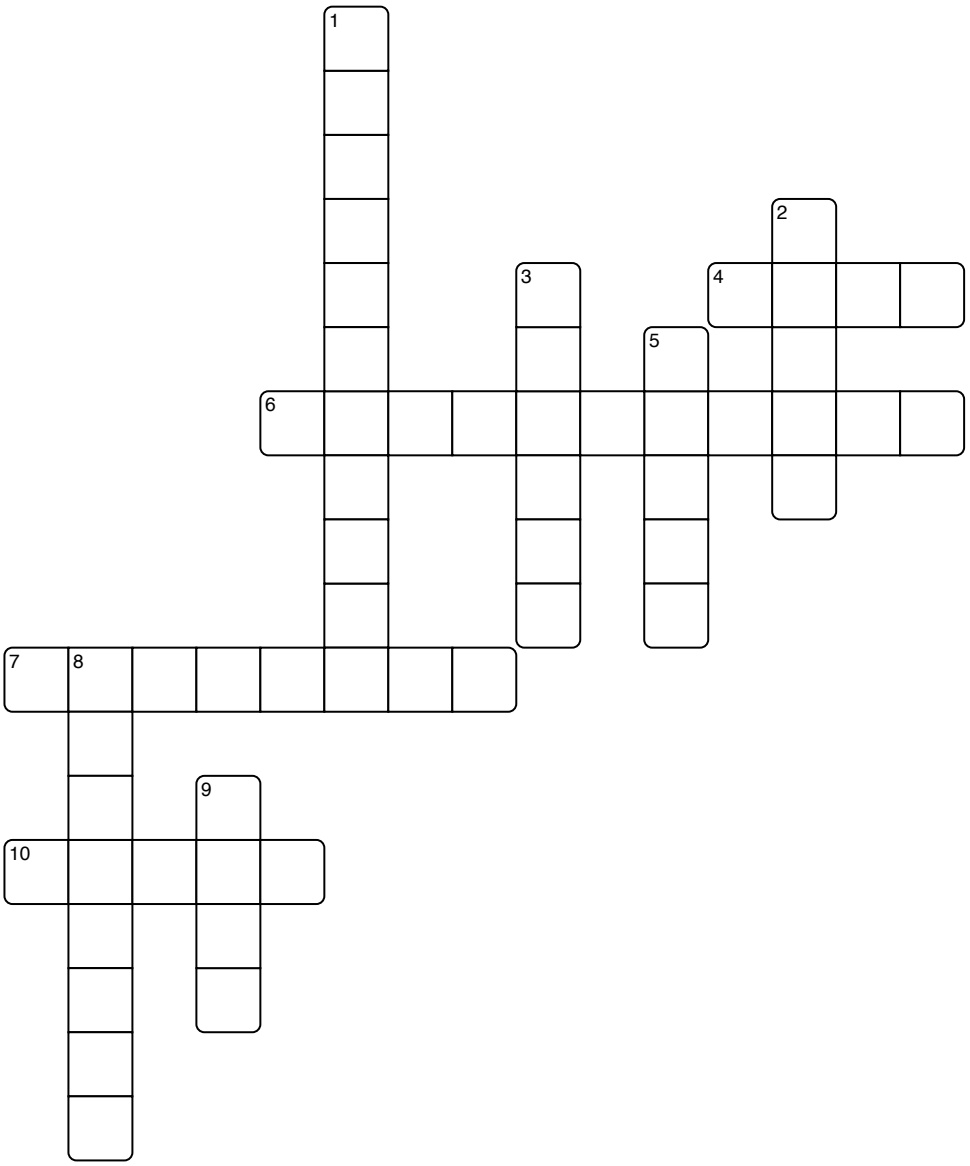
Bonding



Across

4. These are formed when atoms gain or lose electrons. It is better (+) to give than to receive(-)!
6. Ionic compounds are excellent at conducting this.
7. The type of bonding in methane.
10. This is the type of bonding in copper (II) Flouride. Hint: periodic table are the elements-left or right of step?

Down

1. Ionic compounds have this type of structure usually.
2. Ionic compounds are usually found in this state.
3. Ionic bonds are usually formed between_____ and non-_____
5. Non-metals are on the_____side of the step in the periodic table.
8. An ionic bound is the electrostatic force of attraction between _____ charged ions.
9. Ionic compounds have this boiling point.

by Ms. Derwin